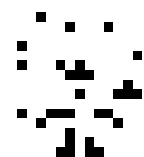




# THE HISTORY OF THE WORLD



CHAPTER I  
THE BEGINNING

THE HISTORY OF THE WORLD  
 FROM THE BEGINNING TO THE PRESENT  
 BY  
 HENRY ADAMS  
 VOLUME I  
 THE BEGINNING TO THE PRESENT  
 PART I  
 THE BEGINNING TO THE PRESENT  
 CHAPTER I  
 THE BEGINNING



THE HISTORY OF THE WORLD

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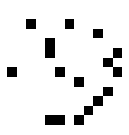
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STATE OF CALIFORNIA  
COUNTY OF LOS ANGELES

1917

Know all men by these presents, that I, the undersigned, do hereby certify that the following is a true and correct copy of the original as the same appears in the records of the County of Los Angeles, California, to-wit:

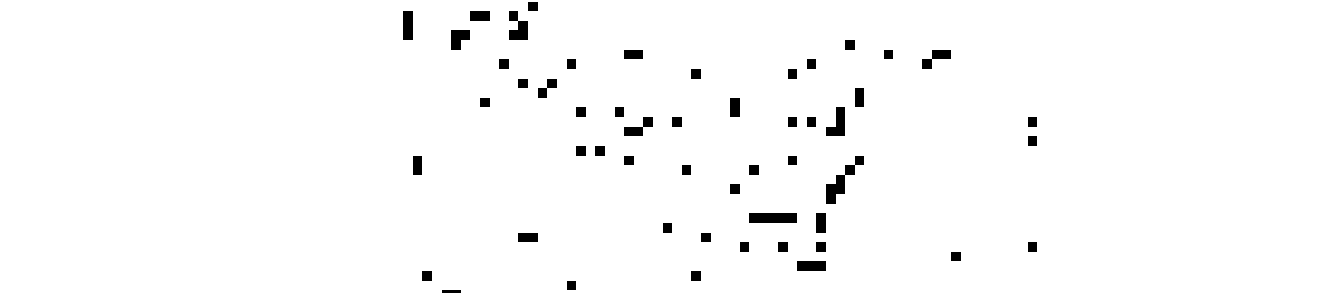
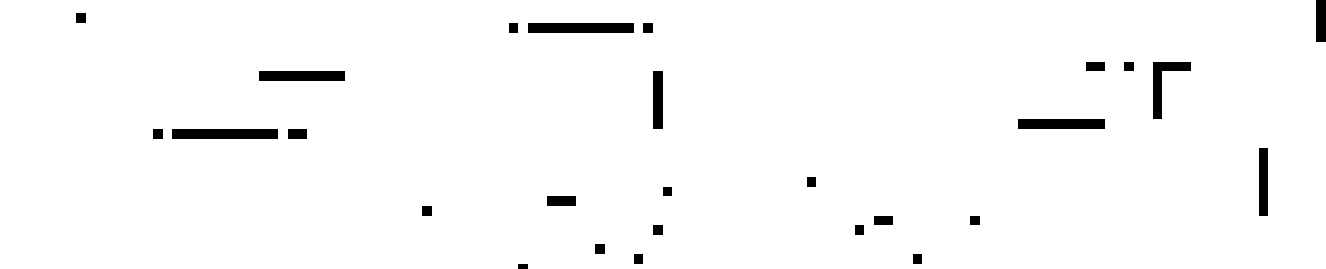
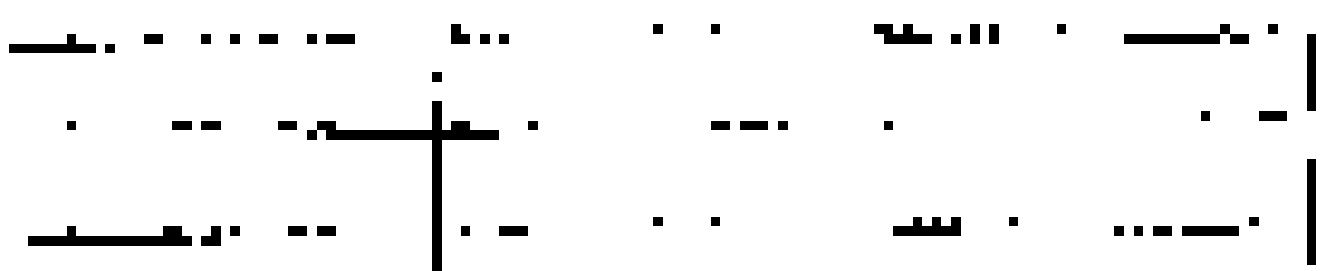
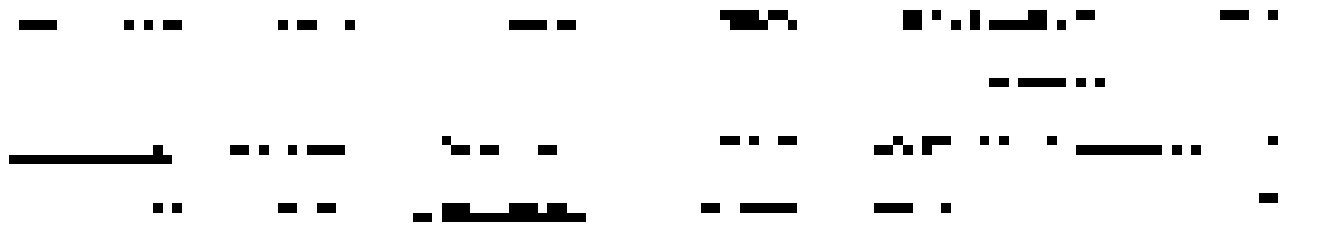
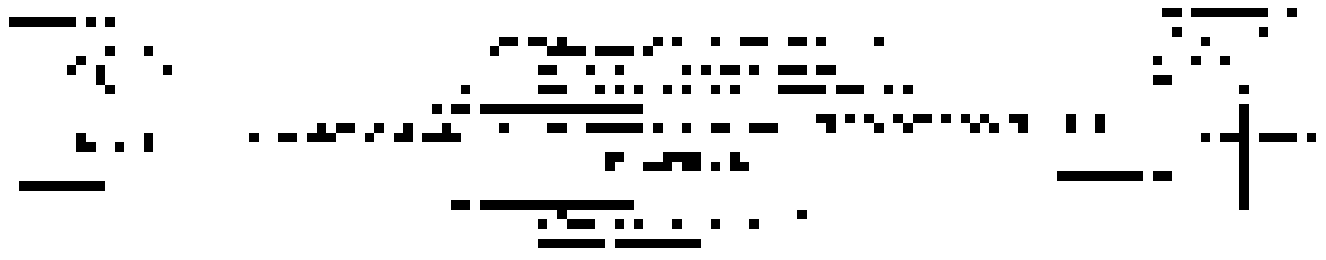
THE RECORDS OF THE COUNTY OF LOS ANGELES, CALIFORNIA, IN THE MATTER OF THE ESTATE OF JAMES EARL RAY, DECEASED.

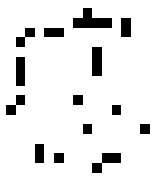
IN WITNESS WHEREOF, I have hereunto set my hand and the seal of said County, at Los Angeles, California, this 15th day of January, 1917.

Notary Public in and for the State of California, My Commission Expires \_\_\_\_\_

\_\_\_\_\_  
Notary Public

STATE OF CALIFORNIA  
COUNTY OF LOS ANGELES





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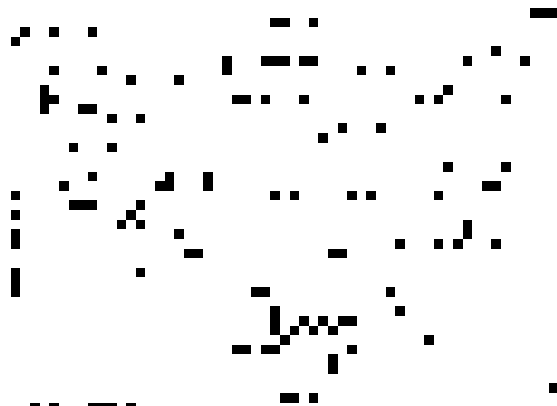
THE UNIVERSITY OF CHICAGO  
DEPARTMENT OF CHEMISTRY

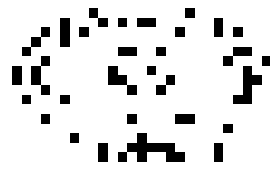
PHYSICAL CHEMISTRY  
PROFESSOR J. H. GOLDSTEIN  
LECTURE 10: THE THEORY OF SOLUTIONS  
PART I: IDEAL SOLUTIONS

DATE: \_\_\_\_\_  
NAME: \_\_\_\_\_

1. Consider a binary mixture of two liquids, A and B, which form an ideal solution. The chemical potentials of the components in the solution are given by:  
$$\mu_A = \mu_A^* + RT \ln x_A$$
$$\mu_B = \mu_B^* + RT \ln x_B$$
where  $\mu_A^*$  and  $\mu_B^*$  are the chemical potentials of the pure components, and  $x_A$  and  $x_B$  are their respective mole fractions in the solution.

Property	Component A	Component B
Chemical Potential	$\mu_A = \mu_A^* + RT \ln x_A$	$\mu_B = \mu_B^* + RT \ln x_B$
Partial Molar Volume	$V_A = V_A^*$	$V_B = V_B^*$
Partial Molar Enthalpy	$H_A = H_A^*$	$H_B = H_B^*$
Partial Molar Entropy	$S_A = S_A^* - R \ln x_A$	$S_B = S_B^* - R \ln x_B$
Partial Molar Heat Capacity	$C_{p,A} = C_{p,A}^*$	$C_{p,B} = C_{p,B}^*$
Partial Molar Gibbs Free Energy	$G_A = \mu_A$	$G_B = \mu_B$





1. 關於本會之組織及職權  
 2. 關於本會之經費及財產  
 3. 關於本會之業務及活動  
 4. 關於本會之紀律及懲戒  
 5. 關於本會之其他事項

中華民國  
 年 月 日

關於本會之組織及職權

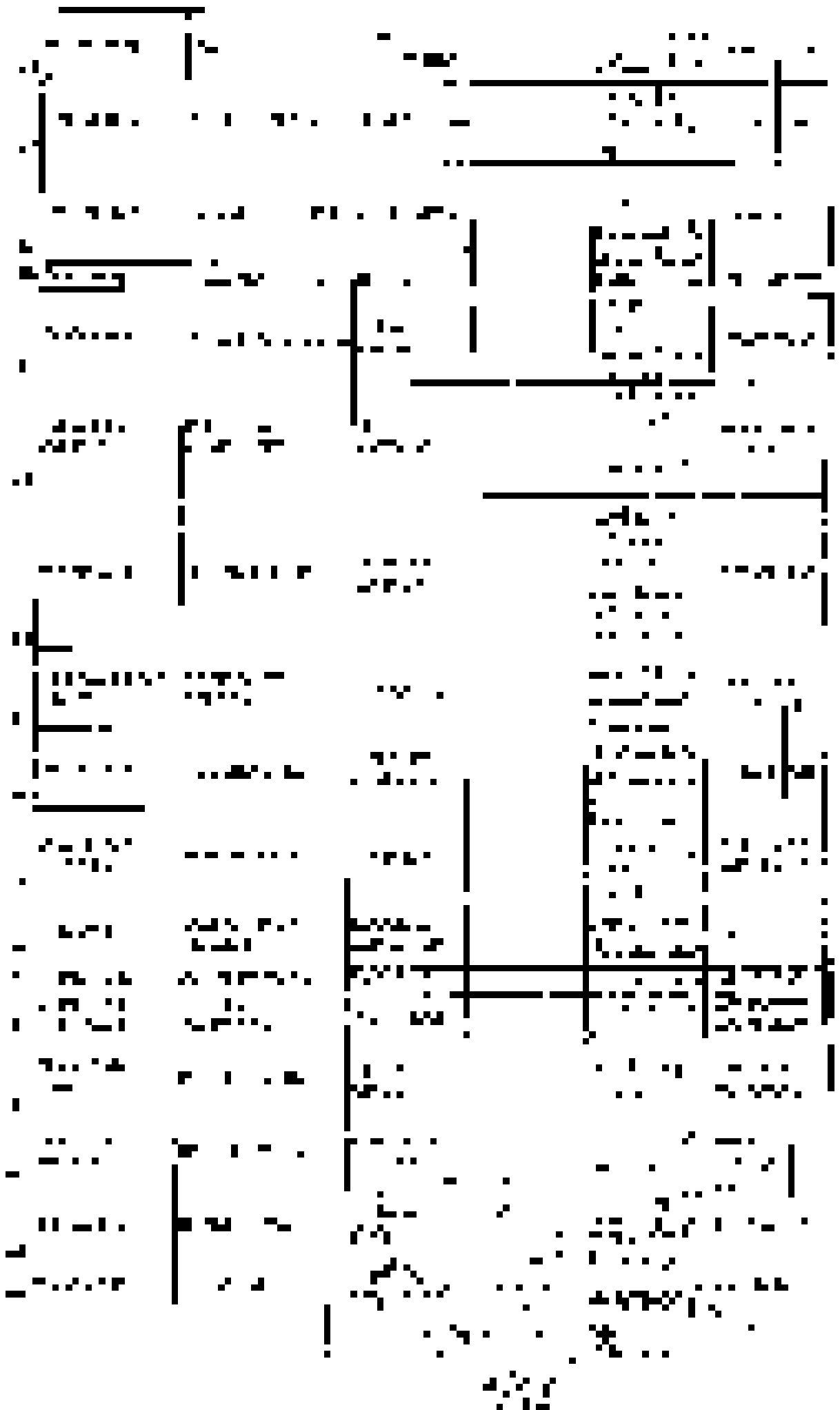
本會之組織及職權如下：  
 一、本會之組織由下列各部分組成：  
 1. 總務部  
 2. 學術部  
 3. 社會服務部  
 4. 紀律委員會  
 5. 其他必要之部門  
 二、本會之職權如下：  
 1. 制定及修改本會之章程  
 2. 審核及批准本會之預算及決算  
 3. 監督及考核本會各部門之工作  
 4. 處理本會之紀律及懲戒事宜  
 5. 辦理本會之其他事務

三、本會之經費及財產如下：  
 1. 本會之經費來源包括：  
 (1) 會員會費  
 (2) 社會捐助  
 (3) 政府補助  
 (4) 其他合法收入  
 2. 本會之財產包括：  
 (1) 不動產  
 (2) 動產  
 (3) 其他財產  
 四、本會之業務及活動如下：  
 1. 學術研究及交流  
 2. 社會服務及公益活動  
 3. 紀律教育及培訓  
 4. 其他必要之業務及活動

五、本會之紀律及懲戒如下：  
 1. 本會之紀律包括：  
 (1) 遵守本會章程  
 (2) 遵守本會各項規章制度  
 (3) 尊重他人權利  
 (4) 維護本會聲譽  
 2. 本會之懲戒包括：  
 (1) 警告  
 (2) 停職  
 (3) 開除會籍  
 (4) 其他必要之懲戒

六、本會之其他事項如下：  
 1. 本會之組織及職權之修改  
 2. 本會之經費及財產之管理  
 3. 本會之業務及活動之調整  
 4. 本會之紀律及懲戒之執行  
 5. 本會之其他必要事項





Year	1950	1951	1952	1953	1954
Production (in thousands tons)	100	100	100	100	100
Consumption (in thousands tons)	100	100	100	100	100
Exports (in thousands tons)	0	0	0	0	0
Imports (in thousands tons)	0	0	0	0	0
Total (in thousands tons)	100	100	100	100	100



THE UNIVERSITY OF CHICAGO  
DEPARTMENT OF CHEMISTRY  
5800 S. UNIVERSITY AVENUE, CHICAGO, ILL. 60637

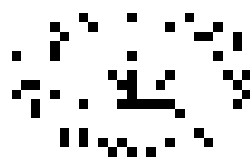
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1	John Doe	123 Main St	Chicago	Ill.	60601
2	Jane Smith	456 Elm St	Chicago	Ill.	60602
3	Bob Johnson	789 Oak St	Chicago	Ill.	60603
4	Alice Brown	101 Pine St	Chicago	Ill.	60604
5	Charlie White	202 Cedar St	Chicago	Ill.	60605
6	Diana Green	303 Birch St	Chicago	Ill.	60606
7	Frank Black	404 Spruce St	Chicago	Ill.	60607
8	Grace King	505 Maple St	Chicago	Ill.	60608
9	Henry Lee	606 Elm St	Chicago	Ill.	60609
10	Ivy Hill	707 Oak St	Chicago	Ill.	60610

PREPARED BY: \_\_\_\_\_  
CHECKED BY: \_\_\_\_\_

UNIVERSITY OF CHICAGO  
DEPARTMENT OF CHEMISTRY  
5800 S. UNIVERSITY AVENUE, CHICAGO, ILL. 60637



# STATE OF TEXAS

## COMMISSIONERS OF THE GENERAL LAND OFFICE

REPORT OF THE COMMISSIONERS OF THE GENERAL LAND OFFICE

FOR THE YEAR ENDING DECEMBER 31, 1900

### CONTENTS

Introduction ..... 1

Statement of the Land Office for the year ending December 31, 1900 ..... 1

Statement of the Land Office for the year ending December 31, 1899 ..... 1

Statement of the Land Office for the year ending December 31, 1898 ..... 1

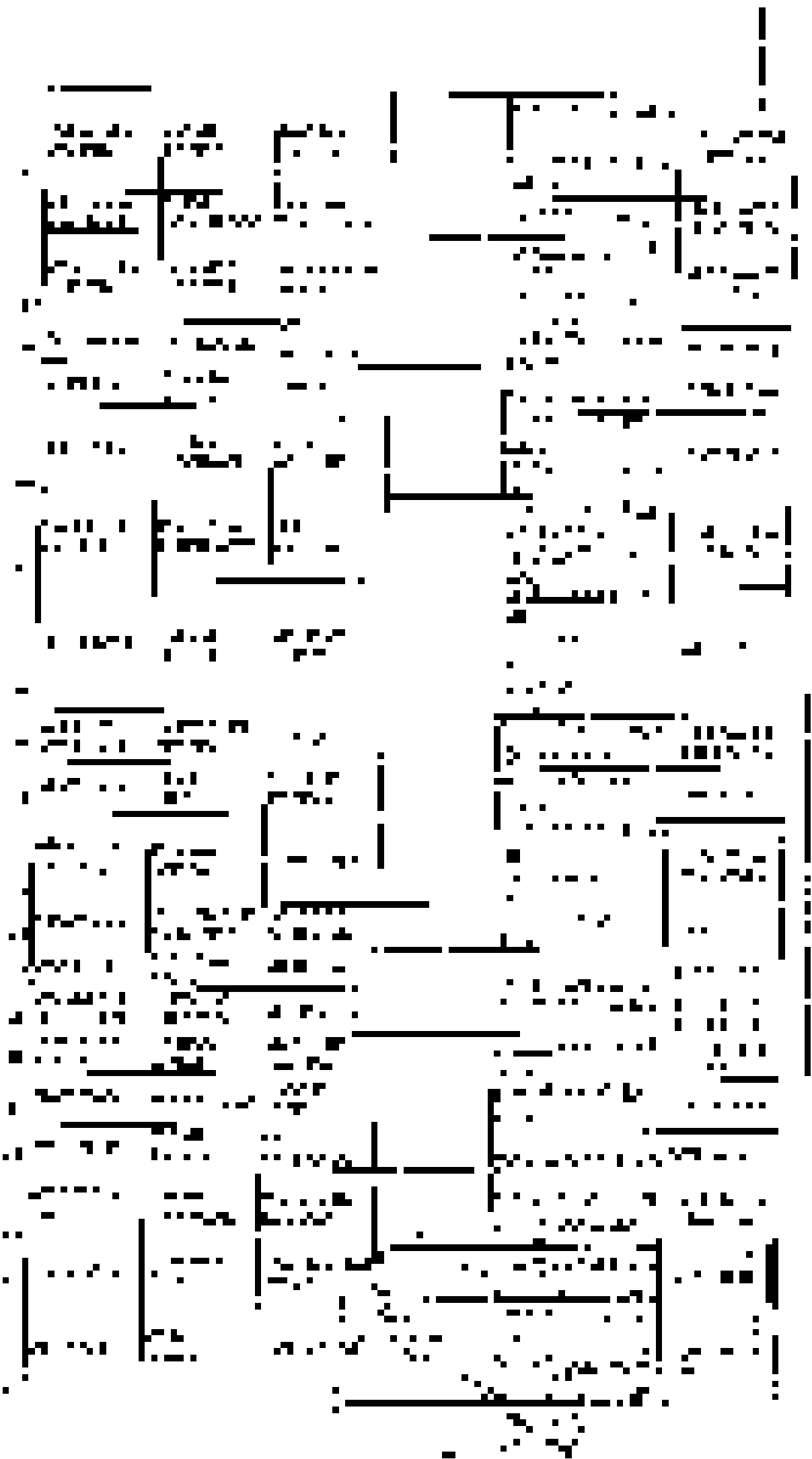
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Statement of the Land Office for the year ending December 31, 1896 ..... 1

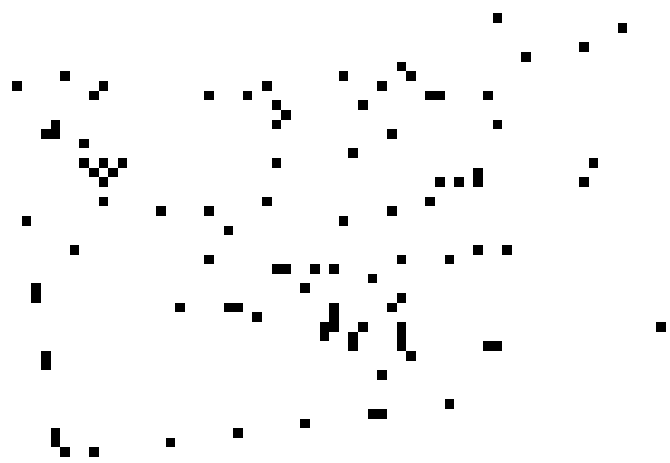
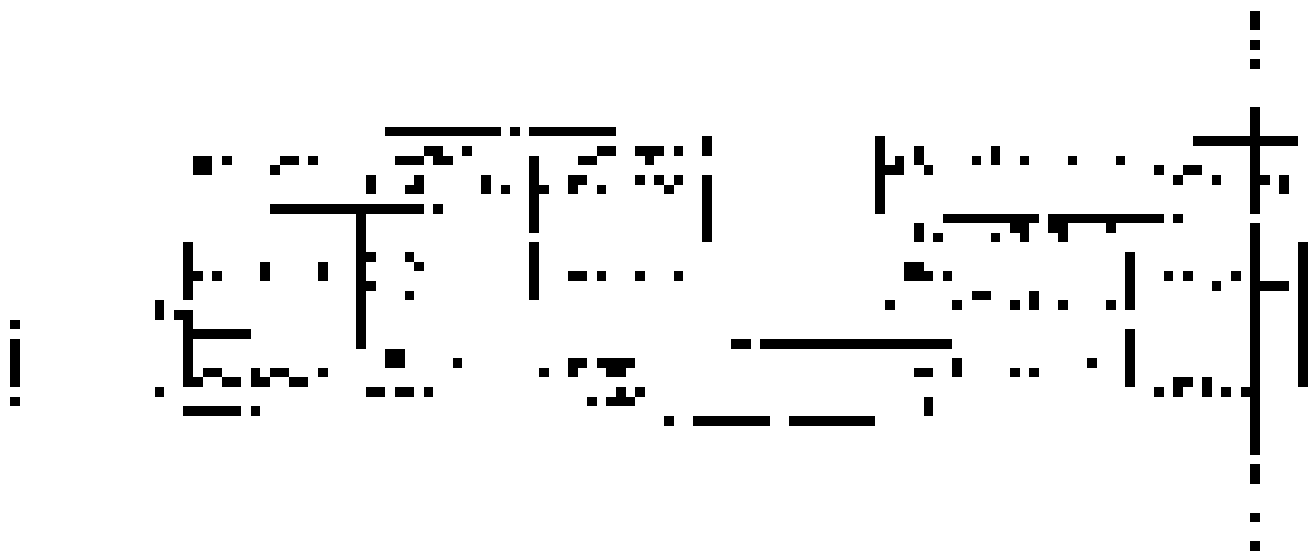
Statement of the Land Office for the year ending December 31, 1895 ..... 1

Statement of the Land Office for the year ending December 31, 1894 ..... 1

Statement of the Land Office for the year ending December 31, 1893 ..... 1

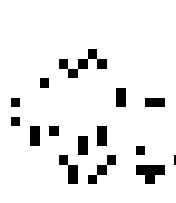






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REPUBLIC OF TURKEY  
 MINISTRY OF NATIONAL EDUCATION  
 NATIONAL CURRICULUM FRAMEWORK FOR  
 PRIMARY EDUCATION

APPENDIX 1

Grade	Subject	Learning Objectives	Assessment Methods	Resources
1st	Mathematics	Identify and count objects, understand basic addition and subtraction.	Observation, oral questioning, written tests.	Mathematics textbooks, manipulatives.
1st	Science	Observe and describe natural phenomena, understand basic properties of matter.	Observation, practical activities, oral questioning.	Science textbooks, laboratory equipment.
1st	Language	Read and understand simple texts, write short paragraphs.	Oral questioning, reading tests, writing tests.	Language textbooks, storybooks.
1st	Arts	Express ideas and feelings through drawing and music.	Observation, portfolio assessment.	Arts textbooks, drawing materials, musical instruments.
1st	Physical Education	Engage in physical activities, develop basic motor skills.	Observation, participation.	Physical Education textbooks, sports equipment.
1st	Social Studies	Understand basic social relationships, identify community members.	Oral questioning, role-playing.	Social Studies textbooks, community maps.
1st	Foreign Language	Learn basic vocabulary and simple sentences in English.	Oral questioning, listening tests.	Foreign Language textbooks, audio materials.
1st	Religion	Understand basic religious concepts, practice good habits.	Oral questioning, practical activities.	Religion textbooks, religious stories.
1st	Health	Understand basic health concepts, practice healthy habits.	Oral questioning, practical activities.	Health textbooks, health posters.
1st	Environmental Studies	Understand basic environmental concepts, practice environmental protection.	Oral questioning, practical activities.	Environmental Studies textbooks, environmental maps.
1st	Information Technology	Use basic computer skills, understand digital literacy.	Oral questioning, practical activities.	Information Technology textbooks, computers.
1st	History	Understand basic historical events, identify historical figures.	Oral questioning, reading tests.	History textbooks, historical maps.
1st	Geography	Understand basic geographical concepts, identify geographical features.	Oral questioning, practical activities.	Geography textbooks, geographical maps.
1st	Local Studies	Understand basic local history and culture, identify local landmarks.	Oral questioning, practical activities.	Local Studies textbooks, local maps.
1st	Character and Citizenship Education	Develop good character traits, understand basic citizenship rights and responsibilities.	Oral questioning, practical activities.	Character and Citizenship Education textbooks, character stories.

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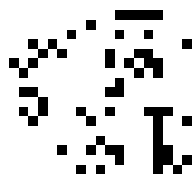
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